

Synthesis of 3,5-difunctionalized 1-methyl-1H-pyrazolo[3,4-*b*]pyridines involving palladium-mediated coupling reactions

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Abstract—Indirect iodination of 2-chloro-nicotinonitrile gave 2-chloro-5-iodonicotinonitrile, which was cyclized with methylhydrazine to lead to 3-amino-5-iodopyrazolo[3,4-*b*]pyridine. Position 3 was then protected by pivaloyl group and the resulting 5-iodo-3-pivaloylamino-pyrazolo[3,4-*b*]pyridine was engaged in palladium-promoted coupling reactions with various reagents to give 3-pivaloylamino-5-substituted compounds. Deprotection and iododediazotiation followed by cross-coupling reactions in position 3 afforded novel unsymmetrical 3,5-disubstituted pyrazolo[3,4-*b*]pyridine species.

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Indoles and aza-indoles are known to be natural or not^{1–8} pharmacophoric moieties represented in many active compounds.

Fused pyrazoles show a wide variety of biological properties.^{9–11} However, few publications are devoted to chemistry of pyrazolo[3,4-*b*]pyridines^{12,13} as indole or indazole isosteres. Particularly, functionalization of pyrazolo[3,4-*b*]pyridines in position 5 was only recently investigated.^{14,15}

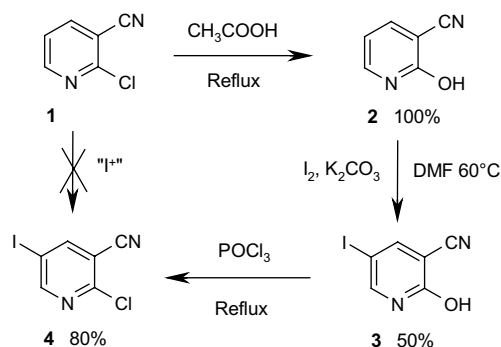
Aiming to extend pyrazolo[3,4-*b*]pyridine libraries, we report here a pathway leading to 3,5-disubstituted-pyrazolo[3,4-*b*]pyridines via successive palladium-catalyzed coupling reactions.

The key intermediate 2-chloro-5-iodonicotinonitrile **4** was prepared by indirect iodination of 2-chloronicotinonitrile **1**. Indeed, the latter was completely unreactive toward electrophilic substitution with an 'I⁺' equivalent such as *N*-iodosuccinimide.

2-Chloronicotinonitrile **1** was transformed by nonanhydrous acetic acid¹⁶ into 2-hydroxynicotinonitrile **2**. The latter was halogenated in position 5 by iodine and potassium carbonate in DMF. Resulting 2-hydroxy-5-iodonicotinonitrile **3** was refluxed with phosphorus

oxychloride to lead 2-chloro-5-iodonicotinonitrile **4** in fair yield (Scheme 1).

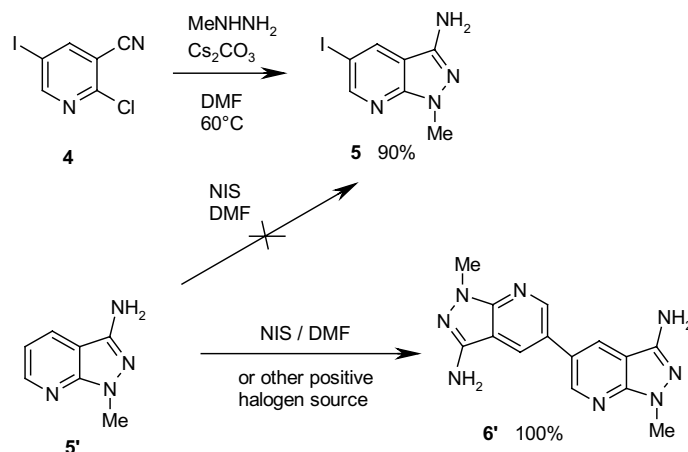
Efficient cyclization of 2-chloronicotinonitrile with various hydrazines using copper iodide and *ortho*-phenantroline as catalyst in the presence of cesium carbonate in DMF was reported earlier.¹⁷ Reaction of 2-chloro-5-iodo nicotinonitrile **4** in these conditions only resulted in degradation. However, cyclization to 3-amino-5-iodopyrazolo[3,4-*b*]pyridine **5** occurred in the absence of catalyst. Direct iodination of known 3-amino-pyrazolo[3,4-*b*]pyridine^{11,17} **5'** only granted dimer **6'** (Scheme 2). This kind of aromatic dimerization or coupling has been already described when reagent like PIFA was used.¹⁸



Scheme 1. Preparation of 2-chloro-5-iodonicotinonitrile.

Keywords: Pyrazolo[3,4-*b*]pyridines.

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Scheme 2. Synthesis of 1-methyl-3-amino-5-iodopyrazolo[3,4-*b*]pyridine.

Prior to cross-coupling pivaloylation¹⁹ of **5** was compulsory to ensure cleaner reaction and easier isolation step (Scheme 3).

Compounds **7a–d** were prepared from 5-iodo-3-pivaloylamino pyrazolo[3,4-*b*]pyridine **6** following Suzuki,²⁰ Sonogashira,²¹ Heck,²² or Stille²³ conditions. Deprotection was performed in anhydrous methanol/hydrogen chloride solution²⁴ to give 1-methyl-3-amino-5-substituted pyrazolo[3,4-*b*]pyridines **8a–d** (Scheme 4).

The transformations proceeded under quite mild conditions and reaction times were often short. The experiments led to the expected compounds in good to excellent yields (Table 1).

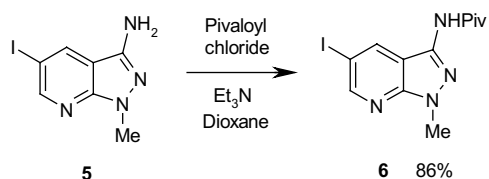
Treatment of 3-amino-5-substituted pyrazolo[3,4-*b*]pyridines **8a** and **8b** by sodium nitrite²⁵ in aqueous acidic

conditions at 0°C led to the corresponding diazonium salt, which was decomposed by potassium iodide²⁶ into 1-methyl-3-iodo-5-substituted pyrazolo[3,4-*b*]pyridines **9a** and **9b** (Scheme 5) in moderate to satisfying yields.

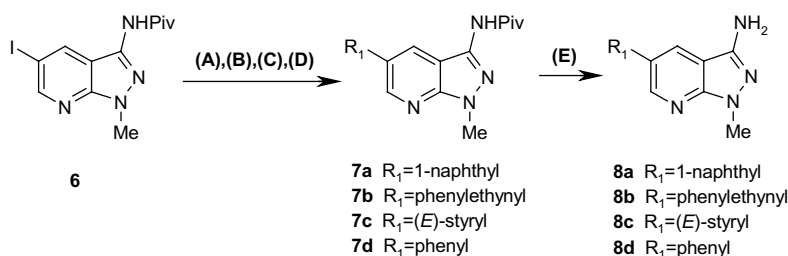
3,5-Difunctionalized compounds **10a,b** were prepared from 1-methyl-3-iodo-5-substituted pyrazolo[3,4-*b*]pyridines **9a,b** according to Sonogashira²⁷ and Heck²⁸ methods (Scheme 6).

Cross-coupling reactions afforded unsymmetrical products in fair yields as shown in Table 2.

In conclusion, we synthesized various pyrazolo[3,4-*b*]pyridines by the mean of easy and clean reactions. The first step consisted in introduction of iodine in position 5 followed by palladium-mediated cross-coupling reactions. A further derivatization using iododediazotiation in position 3 was performed also by palladium-mediated



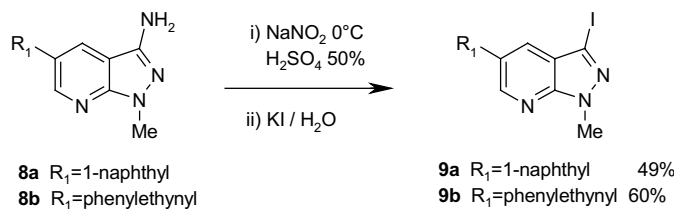
Scheme 3. Pivaloylation of 1-methyl-3-amino-5-iodopyrazolo[3,4-*b*]pyridine.



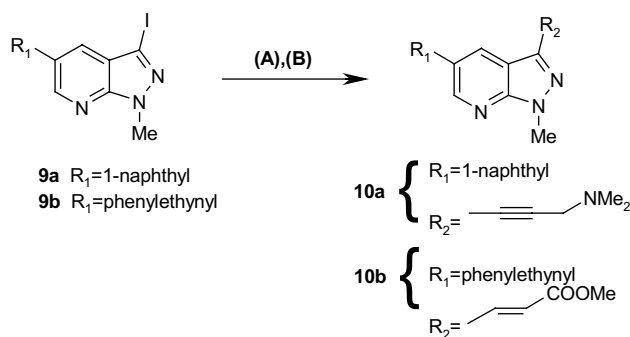
Scheme 4. Conditions and reagents: (A) (naphth-1-yl)-boronic acid, Pd(PPh₃)₄, Na₂CO₃, DME, 75°C, 1h 30 min; (B) phenylacetylene, PdCl₂(PPh₃)₂, CuI, DMF/TEA, rt overnight; (C) styrene, PdCl₂(dppf)₂, *n*-Bu₄NI, DMF/TEA/H₂O, 50°C, 2h; (D) Sn(Ph)₄, Pd(PPh₃)₄, toluene, reflux 2h; (E) MeOH/HCl anhyd reflux overnight.

Table 1. Cross-coupling reactions in position 5 and deprotection of position 3

R ₁	Path	Coupling yield	Deprotection yield
1-Naphthyl	A	7a 90%	8a 93%
Phenylethynyl	B	7b 75%	8b 99%
(<i>E</i>)-Styryl	C	7c 65%	8c 92%
Phenyl	D	7d 50%	8d 94%



Scheme 5. Preparation of 1-methyl-3-iodo-5-substituted pyrazolo[3,4-*b*]pyridines.



Scheme 6. Conditions and reagents: (A) 3-dimethylamino-prop-1-yne, PdCl₂(PPh₃)₂, CuI, DMF/TEA, rt overnight; (B) methyl acrylate, PdCl₂(dppf)₂, *n*-Bu₄NI, DMF/TEA/H₂O, 50 °C, 2h.

Table 2. Synthesis of 1-methyl-3,5-disubstituted pyrazolo[3,4-*b*]pyridines

R ₁	R ₂	Path	Coupling yield
1-Naphthyl		A	10a 63%
Phenylethynyl		B	10b 77%

cross-coupling reactions, affording new dissymmetrical species **10a** and **10b**.

Acknowledgements

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- (C_{arom}): 131.9 (C_{arom}); 137.4 (C₄); 139.1 (C₃); 149.7 (C_{7a}); 152.6 (C₆); 176.7 (C=O); MS: *m/z* = 333 (M + H)⁺; IR (cm⁻¹): 1665 (C=O).
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26. *General procedure for the preparation of 1-methyl-3-iodo-5-substituted pyrazolo[3,4-*b*]pyridines 9a,b*: 3.4 mmol of 1-methyl-3-aminopyrazolo[3,4-*b*]pyridine in 10 mL of sulfuric acid 16N were cooled to 0°C then 3.5 mmol of sodium nitrite in 3 mL of water were added slowly at 0°C. The medium was stirred 1 h at 0°C then 13.5 mmol of potassium iodide in 10 mL of water were poured all at once. The mixture was heated to room temperature during 1 h then brought to pH = 7/8 using solid sodium carbonate and extracted with dichloromethane. The organic phase was washed with a saturated solution of sodium thiosulfate then dried on MgSO₄ and finally evaporated under reduced pressure. The crude was purified on silica gel column chromatography.
- Compound 9b**: 1-methyl-3-iodo-5-phenylethynyl-1H-pyrazolo[3,4-*b*]pyridine: ¹H NMR (CDCl₃): δ_{ppm} 4.16 (s, 3H, CH₃); 7.35–7.38 (m, 3H, H_{arom}); 7.54–7.58 (m, 2H, H_{arom}); 7.93 (d, 1H, H₄, *J* = 1.8 Hz); 8.68 (d, 1H, H₆, *J* = 1.8 Hz); ¹³C NMR (CDCl₃): δ_{ppm} 35.8 (CH₃); 86.3 (C≡CPh); 89.5 (C₃); 91.6 (C≡CPh); 114.2 (C₅); 120.0 (C_{3a}); 122.7 (C_{arom}); 128.6 (C_{arom}); 128.8 (C_{arom}); 131.7 (C_{arom}); 133.0 (C₄); 150.0 (C_{7a}); 150.7 (C₆); MS: *m/z* = 360 (M + H)⁺.
27. See experimental section of Ref. 20 for Sonogashira coupling. **Compound 10a**: 1-methyl-3-(1-dimethylamino-prop-2-ynyl)-5-(naphth-1-yl)-1H-pyrazolo[3,4-*b*]pyridine: ¹H NMR (CDCl₃): δ_{ppm} 2.39 (s, 6H, N(CH₃)₂); 3.58 (s, 2H, CH₂N); 4.24 (s, 3H, NNCH₃); 7.45–7.94 (m, 7H, H_{arom}); 8.21 (d, 1H, H₄, *J* = 1.9 Hz); 8.69 (d, 1H, H₆, *J* = 1.9 Hz); ¹³C NMR (CDCl₃): δ_{ppm} 34.5 (N(CH₃)₂); 44.3 (NNCH₃); 48.7 (CH₂N); 89.1 (C≡CCH₂); 98.0 (C≡CCH₂); 117.0 (C_{3a}); 125.4; 125.5; 126.2; 126.7; 127.2; 128.6; 128.7; 130.4; 130.5; 132.1; 132.2; 133.9; 136.5 (C₄); 149.7 (C_{7a}); 151.1 (C₆); MS: *m/z* = 341 (M + H)⁺.
28. See experimental section of Ref. 21 for Heck coupling. **Compound 10b**: methyl(*E*)-3-(1-methyl-5-phenylethynyl-1H-pyrazolo[3,4-*b*]pyridin-3-yl)-acrylate: ¹H NMR (CDCl₃): δ_{ppm} 3.85 (s, 3H, NCH₃); 4.18 (s, 3H, OCH₃); 6.73 (d, 1H, CH=CHCOOMe, *J*_{trans} = 16.3 Hz); 7.37–7.40 (m, 3H, H_{arom}); 7.56–7.60 (m, 2H, H_{arom}); 7.90 (d, 1H, CH=CHCOOMe, *J*_{trans} = 16.3 Hz); 8.39 (d, 1H, H₄, *J* = 1.8 Hz); 8.70 (d, 1H, H₆, *J* = 1.8 Hz); ¹³C NMR (CDCl₃): δ_{ppm} 34.6 (NCH₃); 52.0 (OCH₃); 86.4 (C≡CPh); 91.9 (C≡CPh); 113.8 (C_{3a} or C₅); 114.7 (C_{3a} or C₅); 120.0 (C=CCOOMe); 122.7 (C_{arom}); 128.6 (C_{arom}); 128.8 (C_{arom}); 131.7 (C₄ or C_{arom}); 132.4 (C₄ or C_{arom}); 135.5 (C=CCOOMe); 139.0 (C₃); 140.5 (C₆); 150.1 (C_{7a}); 151.2 (C₆); 167.2 (C=O); MS: *m/z* = 318 (M + H)⁺; IR (cm⁻¹): 1708 (C=O).